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FAROOK COLLEGE (AUTONOMOUS), KOZHIKODE

Second Semester B.Sc. Degree Examination, March/April 2021 BCH2C02 - Physical Chemistry

(2020 Admission onwards)

Time: 2 hours

Max. Marks: 60

Section A (Short answers)

(Answer questions up to 20 marks. Each question carries 2 marks)

- 1. Under what conditions does a real gas approach ideal behaviour?
- 2. Define an isolated system. Give an example.
- 3. Calculate the R.M.S. velocity of O2 molecule at 27° C.
- 4. Sketch the (200) planes of a face-centred cubic lattice.
- The heat of reaction at constant volume is for the reaction, CH₄ (g) + 2O₂ (g) →CO₂ (g) is 75.83 kJ, at 300 K. Calculate the heat of reaction at constant pressure, at 300 K.
- 6. State Henry's law. Give the unit of Henry 's constant.
- 7. Why is aqueous solution of ferric chloride acidic?
- 8. One mole of water at 373 K changes to steam by absorbing 40.9 kJ of heat. If the work done by the system is 3.5 kJ, calculate the increase in internal energy.
- 9. Write the kinetic gas equation and explain the terms.
- 10. Calculate the degree of ionisation of NH₄OH in 0.02 M solution, the ionisation constant being 1.8X10⁻⁵ mol L⁻¹ at 25⁰C.
- 11. How is entropy related to the heat exchanged reversibly in a process at constant temperature?
- 12. Calculate the concentration of an aqueous solution of a non-volatile solute which exerts an osmotic pressure of 3.731 atm at 300 K.

[Ceiling of marks: 20]

Section B (Paragraph)

(Answer questions up to 30 marks. Each question carries 5 marks)

- 13. (a) State and explain the third law of thermodynamics.
 - (b)Calculate the Gibb's free energy change at 25°C for the reaction,
- . $CO_{(g)} + Cl_{2(g)} \rightarrow COCl_{2(g)}$. Given, $\Delta H = -109 \text{ kJ}$ and $\Delta S = -137 \text{ JK}^{-1}$. Predict whether the reaction is spontaneous or not.
- 14. (a) Derive an equation relating the enthalpy change and internal energy of a reaction.(b) Distinguish between isothermal and adiabatic process.
- 15. Give the principle of conductometric titrations and discuss the conductometric titration curves of
 - (a)Strong acid against a strong base. (b) weak acid against weak base.
- 16. What is meant by an ideal gas? What are the causes of deviation of a real gas from ideal behaviour?
- 17. Explain the effect of dilution in the specific conductance and molar conductance of a strong electrolyte.
- 18. Explain the factors affecting the solubility of a gas in a liquid.
- 19. Discuss the various stoichiometric defects in crystals.

[Ceiling of marks: 30]

Section C (Essay)

(Answer any one. Each question carries 10 marks)

- 20 Derive Bragg equation and discuss its applications.
- (a) What are fuel cells. Discuss the functioning of H₂.O₂ fuel cell
 (b) Write the cell reaction and calculate the EMF of the electrochemical cell,
 Fe | Fe²⁺ (0.1 M) | Cd²⁺ (0.001 M) | Cd, at 25° C. Given, E°Fe²⁺/Fe= -0.44 V and Cd²⁺/Cd = -0.40 V

 $[1 \times 10 = 10]$

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FAROOK COLLEGE (AUTONOMOUS), KOZHIKODE

Second Semester B.Sc Degree Examination, March/April 2021

BMT2C02 - Mathematics - 2

(2020 Admission onwards)

Time: 2 hours

Max. Marks: 60

PART A

Answer all questions. Each question carries 2 marks.

Maximum marks from this section is 20.

- 1. Convert the polar coordinate $(4, -\pi)$ into cartesian coordinate.
- 2. Differentiate $\cosh^{-1} x^2$
- 3. Consider the curve $x = 3 \cos t$, $y = \sin t$. Find the points where the tangent is horizontal.
- 4. Find $\lim_{n\to\infty} \frac{3n^2+1}{n^2+n}$
- 5. Evaluate $\int_0^{\frac{\pi}{2}} \cos x \ dx$ by using Trapezoidal rule with n = 4.
- 6. Show that the series $\sum_{i=1}^{\infty} 1 + \frac{1}{2^i}$ diverges.
- 7. Test the convergence of $\sum_{n=1}^{\infty} \frac{3^n}{n^2}$.
- 8. Give an example of a vector space. Explain your answer
- 9. Check whether the set of vectors (3,5), (2,10) are linearly independent or not.
- 10. Find the inverse of the matrix $\begin{bmatrix} 1 & 3 \\ 4 & 10 \end{bmatrix}$
- 11. Verify that the matrix $A = \begin{bmatrix} cost & sint \\ -sint & cost \end{bmatrix}$ is orthogonal.
- 12. Evaluate the determinant of the matrix $A = \begin{bmatrix} 6 & 5 & 0 \\ -1 & 8 & -7 \\ -2 & 4 & 0 \end{bmatrix}$

PART B

Answer all questions. Each question carries 5 marks

Maximum mark from this section is 30

- 13. Show that $\sinh^2 x = \frac{\cosh 2x 1}{2}$
- 14. Find the length of the curve $f(x) = (x-1)^{\frac{3}{2}} + 2$ on [1, 2].

- 15. (a) Write Alternating series test.
 - (b) Show that the series $1 \frac{1}{2} + \frac{1}{3} \frac{1}{4} + \frac{1}{5} \cdots$ converges.
- 16. Find the Taylor series generated by $f(x) = e^x$ at $x_0 = 0$.
- 17. Let $B = \{u_1, u_2\}$, where $u_1 = (3,1), u_2 = (1,1)$ is a basis for R^2 . Find an orth basis for R^2 using the Gram Schmidt orthonormalization process.
- 18. Solve the linear system

$$x_1 + 2x_2 - x_3 = 0$$
$$2x_1 + x_2 + 2x_3 = 9$$
$$x_1 - x_2 + x_3 = 3$$

Using Gaussian elimination.

19. Find the rank of the matrix $\begin{bmatrix} 1 & 1 & 1 \\ 1 & 0 & 4 \\ 1 & 4 & 1 \end{bmatrix}$

PART C

Answer any ONE question. One question carries 10 marks

- 20. (a) Graph the polar curve $r = \cos 2\theta$.
 - (b) Find the area enclosed by the cardioids, $r = 1 + \cos \theta$, $0 \le \theta \le 2\pi$.
- 21. Find the eigen values and corresponding eigen vectors of the matrix $A = \begin{bmatrix} 7 \\ 3 \end{bmatrix}$.

 Also verify Cayley Hamilton theorem.

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FAROOK COLLEGE (AUTONOMOUS), KOZHIKODE

Second Semester B.Sc Degree Examination, March/April 2021 BPH2B02 – Mechanics - II

(2020 Admission onwards)

Time: 2 hours

Max. Marks: 60

The symbols used in this question papers have their usual meanings

Section A- Short Answer Type

(Answer all questions in two or three sentences, each correct answer carries a maximum of 2 marks)

- 1. A simple pendulum hangs in a car that has acceleration 'a'. What is the angle that the string of the pendulum makes with the vertical?
- 2. What are the two fictitious forces that act on object that moves with respect to a rotating frame?
- 3. Write down the expression for the time for the plane of oscillation to rotate once for a Foucault pendulum at latitude λ .
- 4. Give four general properties of central force motion?
- 5. What is meant by reduced mass? What is its advantage?
- 6. Give two advantages and disadvantages of resonance.
- 7. What is satellite maneuvering? Why is it required?
- 8. Write down the differential equation of a damped harmonic oscillator. What is its solution?
- Write down the expressions for maximum kinetic energy, maximum potential energy, total
 energy and average total energy of a particle executing simple harmonic motion. Given that the
 displacement x = A cos(ω₀t).
- 10. What are the two minimum requirements of a medium to support mechanical wave propagation?
- 11. What is the difference between phase velocity and group velocity?
- 12. What is the unit of sound intensity level? How is it defined?

(Ceiling-20)

Section B- Paragraph/ Problem Type (Answer all questions in a paragraph of about half a page to one page, each correct answer carries a maximum of 5 marks)

- Calculate the deflection of a stone dropped from a height of 50m at the equator due t force.
- 14. Draw the variation of U_{eff} with separation r in the energy diagram. Show the variou planetary orbits with different total energies.
- 15. Derive the dispersion relation for a longitudinal waves on a beaded spring with spring K, bead mass m and separation between the beads a.
- 16. A musician's tuning fork rings at 440Hz. The intensity of the sound produced decrea fifth of its original intensity in 4 s. What is the Q factor of the tuning fork?
- 17. Halley's comet is in an elliptic orbit about the sun. The eccentricity of the orbit is period is 76 years. The mass of the sun is 2×10^{30} kg, $G = 6.67\times10^{-11}$ Nm²/kg². Cal distance of perihelion and aphelion from the sun for this comet.
- 18. If the dispersion relation is $\omega^2 = \omega_p^2 + c^2 k^2$, obtain the phase velocity v_p .
- 19. Describe the cause of sea tides due to earth's revolution around the sun.

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Section C- Essay Type (Answer any one question. Answer carries 10 marks)

- Assuming the Fourier series expansion is valid for a periodic function F(t), obtain the
 integral representation of a nonperiodic function ψ(t).
- 21. Describe the equation of a damped harmonic oscillator. Assuming the solution displacement explain the time dependence of amplitude and energy of oscillation a factor.